

Thyroid Protector Potassium Iodide Capsules, Dietary Supplement Compound for Hypothyroidism

140 customer reviews | 8 answered questions







Price: \$13.48 (\$0.22 / count) & FREE Shipping

 $\textbf{Note:} \ \mathsf{Not} \ \mathsf{eligible} \ \mathsf{for} \ \mathsf{Amazon} \ \mathsf{Prime}.$

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About the product

- PILLS MADE IN THE USA: This potassium iodide compound by Performance Supplement Store is manufactured in the United States for optimal quality. These capsules are designed to work as a treatment for hypothyroidism, or an underactive thyroid, but they are also popular among preppers and make a great addition to survival kits.
- BEST PROTECTION FOR HYPOTHYROIDISM OR PREPPERS: For any doomsday preppe

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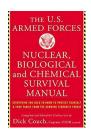


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Product description

For anyone suffering from hypothyroidism, there are a wide variety of treatment options available. However, many of these are high priced or are otherwise inconvenient. Fortunately, this potassium iodide compound by Performance Supplement Store is a natural and affordable dietary supplement that helps to regulate thyroid function, helping with hypothyroidism or an underactive thyroid. This compound comes in convenient capsule form and can easily be taken along with food as part of your daily diet for increased health and peace of mind as you tackle the challenges life throws at you. Unlike many other compounds, pills, or tablets on the market, this Performance Supplement Store treatment is designed for optimal absorption speed and efficiency. Since it's quick and easy for this potassium iodide compound to work its way into your system, you will experience the helpful effects of the supplement faster, and the effects will be more beneficial to your overall health. This supplement is formulated for both men and women and can be taken daily to supplement your health and vitality. This compound is also great for doomsday preppers as part of a survival kit. Potassium iodide is an effective defense for all of your prepper needs. By including it in your kit, you survival enthusiasts and doomsday preppers can find great peace of mind in the face of an emergency. This supplement ships promptly—usually within 24 hours of the initial order placement—so you can start living a healthier life as soon as possible.

Product details

Item Weight: 2.08 ounces

Shipping Weight: 2.9 ounces (View shipping rates and policies)

ASIN: B00N0INFAG **UPC:** 852662693843

Average Customer Review: 140 customer reviews

Amazon Best Sellers Rank: #80,976 in Health & Household (See Top 100 in Health & Household)

#69 in Health & Household > Vitamins & Dietary Supplements > Minerals > Potassium > Potassium Iodide

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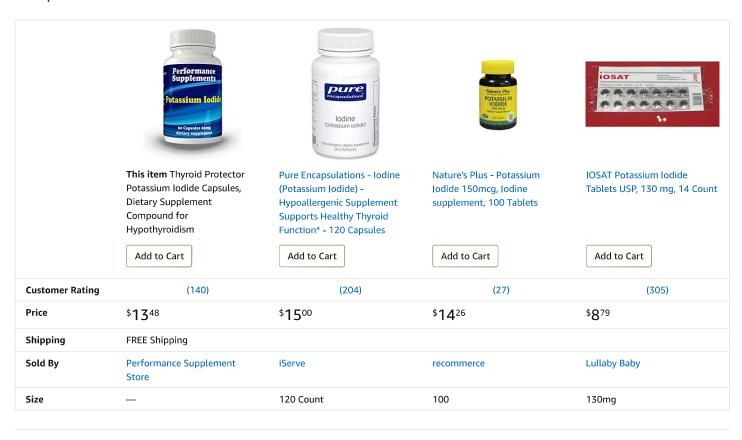
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Ingredients

Potassium Iodide, Rice Flour

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Customer questions & answers

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2 votes Ouestion: Can this be used for a catalyst for rocket fuel

I think you would be looking for potassium nitrate not potassium iodide.

By Scott on September 20, 2016

No but it will protect your thyroid from radiation. By william g smith on February 24, 2016

Collapse all answers

2 votes Question: What BRAND are these? Not mentioned on site, or visible in picture, nor is there a record of the UPC.

> No UPC code on bottle. Label has Performance Supplement store and on printed on the bottom the MFG date, in this case 1/2014. The label indicates a Exp. date of 04/2018 and a lot #04232012. It would be my guess that this was not

manufactured in the USA but probably somewhere like India.

Hope this helps.

By DENNIS K. ALTOM on December 23, 2014

See more answers (2)

1 vote What kind of pill are they? Are they capsules, individually wrapped in the bottle or are they free-floating in the bottle?

Answer:

Question:

Capsule floating in the bottle - similar to free floating Tylenol capsule. - and reduced cesium count in half over three days, patient is recovering and responding well.

By William Hayes on October 16, 2016

They are capsules free floating in a bottle By Rosemary on October 17, 2016

Thanks for Asking, There are 60 individually wrapped Potassium iodide Capsules free floating in the plastic container. I hope that helps. Take care. Eric

By Performance Supplement Store SELLER on October 14, 2016

Collapse all answers

0 votes Question: What is the expiration of the current batch?

The current batch was manufactured 4 months ago. With a 5 year on expiration, however we used our old labels that say 9/2021. So by label 2021. Actual 2023. And some say they don't ever expire. Hope that helps. Take care.

By Performance Supplement Store **SELLER** on March 3, 2018

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Customer reviews

140

Answer:

4.6 out of 5 stars

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Top customer reviews



Alex L

Truly is what it claims to be. I proved this independently via a chemical reaction. Also, this stuff doesn't expire.

December 24, 2015

Verified Purchase

Alright, this stuff is the real deal. Each pill really has 65Mg of potassium iodide in it. I can (and did) prove it through a simple science experiment. Long story short, if you react potassium iodide with lead nitrate, you get lead-iodide as a solid precipitate. This settles to the bottom of your beaker while all the other ingredients remain suspended in, free-floating around in the water. So, at this point, drain off the excess water and other reagents, and as long as you know how much potassium iodide you started with (allegedly 65mg, right?) you can prove, mathematically via the chemical equation and molar masses, how much lead nitrate should form. If substantially less (or none at all) forms, then these are fake. In my experiment, I validated that these are *not* fake, but are in fact the real deal.

Before I continue, I should also mention that as a side note, POTASSIUM IODIDE DOESN'T EXPIRE. It's a form of salt. Putting an expiration date on these is stupid. That's like saying that your table salt is going to expire in X years. IT DOESN'T. You could go eat 100 year old table salt and it would be just as good as whats on the shelf at your local grocery store. In fact, your table salt from the grocery store, which was extracted from a salt mine, is usually already thousands of years old. There are still high schools using batches of KI for science experiments that they've hung unto singe the 70s and 80s, I kid you not. The *only* thing that could go wrong is that, as pills, there is a binding agent that holds it all together in a pill form. That being said, if this binding agent ever goes bad, the pills would turn to a flowery dust. So, just give the bottle a shake every 10 years. Still hear pills rattling around instead of a tube of flowery dust? Then there still good. Alright, back to the experiment that proves these are real now.

TO PREFACE WHY I DID THIS: I figured I should prove this to myself since everyone else is just giving it 5 stars without being able to attest if this is *really* what it says it is or not. I mean, I could go sell salt pills, and unless you plan on rubbing nuclear waste all over your body like a pool boy in a bad porno, you really wouldn't know, would you? You'd just give this product 5 stars and say hopefully you'll never have to use it, without knowing if it really works or not (like everyone else is doing!), wouldn't you? Or maybe you're like that one guy who rated it one star out of fear because you don't know how to test it? Hell....maybe you bought a different, "more reputable" product, and hopefully didn't get some fake-knockoff from china that totally isn't even the FDA approved medication it *claims* to be. Well, alright. Here's how I proved that each pill really does have 65mg of potassium iodide in it.

ANYWAY, HERE'S HOW TO REPEAT THE EXPERIMENT, in case anyone else wants to try it, or would like to review my methodology:

*Disclaimer: This experiment uses lead nitrate, which is toxic. If you intend to reproduce this experiment, please exercise caution and follow proper handling procedures for toxic materials. Please dispose of *all* used chemicals responsibly and appropriately. I do not accept any responsibility for any harm you cause to yourself and/or anything and/or anyone else by doing this experiment:

Pb(NO3)2 (Lead Nitrate) in an aqueous solution plus KI (potassium iodide) will form a precipitate. The full equation is as follows: 2KI(aq) + Pb(NO3)2(aq) -> 2KNO3(aq) + PbI(s)

If you're going to reproduce this reaction and don't know what it looks like, look it up on youtube just to verify that all of the visual properties of the reaction match up. (Call me paranoid, but I could name a half dozen other things that will react with lead nitrate and form a precipitate, but none will do so with the proper color and other visual attributes as potassium iodide)

To perform this the way you're "supposed to" you should use the molar mass of the two substances, and a



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Lee

Five Stars

Love it

Published 2 months ago



Jo

Five Stars

awesome

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Ryan Heidelberg

Four Stars

Thank you.

Published 2 months ago



Amazon Customer

Five Stars

lodine is a necessary trace mineral of which most Americans are deficient. Take it to avoid thyroid problems.

Published 3 months ago



Gail McNeil

Five Stars

Excellent-Thank you Published 3 months ago



Aliilani

Five Stars

bought for bug out bag Published 4 months ago



Diadem

It does what is says it does. I can ...

It does what is says it does. I can tell when I take it and when I don't. I haven't had my blood tested with it yet, so can't give an clinical proof.

Published 5 months ago



Amazon Customer

Five Stars

thx

Published 5 months ago

little stoichiometry, to see that you will need about 64.8 milligrams of lead nitrate to react with the 65 milligrams of potassium iodide in a pill. That being said, I added 10% more lead nitrate than I needed, juuuust to make sure the KI was the limiting reagent. Let me also mention now that I used TEN pills in this test. The college level lab I have access to, nor my skills, would be good enough to get a proper, accurate reaction on just 65 milligrams. So, 10 pills, 650 milligrams...*that* I can get an accurate experiment out of. I should also note that the entire experiment was done in a glove box - a tabletop clean room, if you will, to minimize contaminating it with things (like dust) that could throw off my measurements.

I crushed up the pills, and dumped the contents into a 250ml beaker full of of distilled water, dropped in a stirring pellet, and let it go to work for a few hours. Once everything was dissolved, I let it sit for a day to ensure all the "fillers" (I.e. binding agents that actually make the pills take shape!) precipitated to the bottom. Once this was done, I poured the solution that now, should contain only the potassium and iodide ions suspended in the water, into another beaker. The total water ans wet binding agents left at the bottom of the 1st beaker account for about a 1-5% reduction in total anticipated precipitate (lead iodide) from the finished reaction. So, at this point I steamed off the solution to fit in a 50ml beaker. As this weighs much less than a 250ml beaker, it's a better vessle to more accurately get a final reading of total lead nitrate when the experiment is done.

So....blah blah, I added aqueous lead nitrate to aqueous KI from a pill. Precipitate began to form. I let this happen in the 50ml beaker with a stir bar over the course of a couple hours, just to make *damn* sure I really reacted everything. After this was done, I let the precipitate settle overnight. Then, I drained out as much of the water and dissolved lead and potassium nitrates as I could...then I added more distilled water...I did this 4x over two days to ensure that I didn't accidentally remove any precipitate, but that I did remove, most likely upwards of 95% of the nitrates. This is necessary because the whole point of this experiment is to determine the final remaining amount of lead iodide (PbL). If I hadn't done this, the nitrates would have been left behind as well as powdered solids, and I would just have a measure of how many ingredients I have, instead of a *specific* one that proves how much lodide (and therefore Potassium, and therefore entire purpose of this pill!) I started with.

ANYWAY...grand finale folks.

I started with a 50ml beaker that weighs 31011 milligrams, and a stirring pellet that weighed 2254 milligrams. Total weight: 33265 starting milligrams! Post reaction, I ended with a total weight of 34066 milligrams! 33265 - 34066 = 801 Milligrams of lead iodide left behind in my 50ml beaker. The theoretical maximum yield for this experiment would have been 900 milligrams of lead iodide. That means I was able to prove that there is at least 89% (801/900) of the stated 65mg of KI in each pill. That being said, It's probably really around 100, as getting a perfect, 100% reaction, just doesn't happen. Seriously - most high schoolers in AP chem that perform this same reaction in their labs only get a 70-80% reaction due to their own errors and loss of product along the way. So, all things considered, yeah....this is the real deal.

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320 people found this helpful

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Shane Antonio Benitez

Seems great from lab test

May 8, 2017

Verified Purchase

I haven't taken these yet because the nuclear war hasn't started yet but if I survive the fast approaching nuclear attack I make sure I leave a review if these help cure my radiation poison

5 people found this helpful

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Auth

Better to have it and not need it than need it and not have it and then suffer from horrible radiation poisoning.

November 29, 2017

Verified Purchase

Well, I haven't had to use them yet because WW3 keeps getting delayed, but I sleep better knowing I have them just in case.

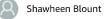
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Billie

Must have to protect your thyroid against radiation posonining



Protect you from radiation

poisoning.

It has a good amount of pills to protect you when need to.

Published 5 months ago



hammy

Five Stars

Man.. I hope they work Published 5 months ago

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