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Re: Gels to use with DMSO

By Guest guest, January 27, 2011 in Health, Medicine and Natural Healing 11

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Guest guest

Posted January 27, 2011



Guests

I use Dr 's DMSO Gel - it is much easier to use than the liquid

Gels to use with DMSO

>I wonder if there are gels that can be used to thicken up DMSO for applying

>on the skin. A few that are easily available are guar gum, acacia gum, and

>xanthan gum. How would one find if the particle size was OK to use? I

>saw the DMSO gel that is for sale and it didn't list the gelling agent.

> I'm also wondering about how to hold gauze on over a gel with DMSO, would

> a non-latex bandaid have toxic material in it?

> Thanks,

> List Home Page:

> http://health.groups.yahoo.com/group/DimethylSulfoxide-DMSO

> Books:

> DMSO Nature's Healer by Morton

> MSM The Definitive Guide by Stanely MD and Appleton, NDYahoo!

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No virus found in this incoming message.

Checked by AVG - www.avg.com

Version: 8.5.449 / Virus Database: 271.1.1/3398 - Release Date: 01/23/11

07:34:00

+

Guest guest

Posted January 28, 2011

Quote



I googled carbopol-

Guests

Thanks, Jane, I looked up Dr. 's DMSO gel, at least it gives some ingredients- carbopol.

"Carbopol polymers are polymers of acrylic acid cross-linked with polyalkenyl ethers or divinyl glycol. They are produced from primary polymer particles of about 0.2 to 6.0 micron average diameter. The flocculated agglomerates cannot be broken into the ultimate particles when produced. Each particle can be viewed as a network structure of polymer chains interconnected via cross-linking. " divinyl glycol? Does that sound desirable to me? No. Guar gum is described sizewise as "Six guar gum samples with a wide range of particle size (70–470 & #956;m) were prepared directly from the same batch of guar seed endosperm " This may make it too large to go through the skin and so a good gel candidate. If I can get 8 oz of DMSO for \$22 why do I want to pay \$46 for 4 oz of gel?

- > I use Dr 's DMSO Gel it is much easier to use than the liquid
- >
- > Jane
- >
- > Gels to use with DMSO
- >
- >
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- >>-----

> > List Home Page: >> http://health.groups.yahoo.com/group/DimethylSulfoxide-DMSO > > Books: > > DMSO Nature's Healer by Morton >> MSM The Definitive Guide by Stanely MD and Appleton, NDYahoo! > > Groups Links > > > No virus found in this incoming message. > Checked by AVG - www.avg.com > Version: 8.5.449 / Virus Database: 271.1.1/3398 - Release Date: 01/23/11 > 07:34:00

Guest guest

N

Guests

Posted January 28, 2011

Quote

I thought a Dr 's product would be OK & it was a gift from a member of

the list

Having used guar gum to thicken patients' fluids I found even in normal $\mbox{\ }$

fluids it was fairly difficult to use - I have no idea how it would be

trying to mix it with DMSO.

Best,

Jane

Thanks, Jane, I looked up Dr. 's DMSO gel, at least it gives some

ingredients- carbopol.

I googled carbopol-

"Carbopol polymers are polymers of acrylic acid cross-linked with polyalkenyl ethers or divinyl glycol. They are produced from primary polymer particles of about 0.2 to 6.0 micron average diameter. The flocculated

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+

Quote

Guest guest



Guests

Posted January 29, 2011

Jane- there may be nothing wrong with carbopol besides the plastic-y sounding names, that article says it is used as a matrix in slow-release pills, and it has a diameter of .2 microns which I figure is 2000 daltons, so too big to be absorbed through the skin according to what I read.

 $\underline{\text{http://www.pharmainfo.net/reviews/carbopol-and-its-pharmaceutical-significance-r}} \\$

eview

I looked up xanthan gum, it also looks too big to go through the skin at 2-4 million daltons

 $\underline{http://www.fda.gov/Food/FoodIngredientsPackaging/GenerallyRecognizedasSafeGRAS/G} \\$

RASListings/ucm153937.htm

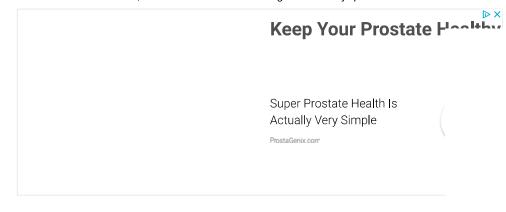
"Ingredients Solutions describes xanthan gum (ethanol precipitate) as an anionic, soluble exopolysaccharide produced by fermentation of the nontoxigenic and nonpathogenic bacterium Xanthomonas campestris. It has a molecular weight of approximately 2-4 million daltons and is composed of a beta-(1-4)-D-glucose backbone with a trisaccharide side chain linked to the third carbon of every other glucose residue."

I didn't know it was a product of bacterial fermentation, I guess I thought it came from seaweed.

Acacia particle size seems harder to track down but may be 580,000 Daltons molecular weight. Actually they all seem good candidates, I will keep researching them.

+

Quote



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