



Global Bromine Industry And Its Outlook

*- Production by Country, Production Process,
Application and Forecast.*

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5. 9. 2008





Content

- 1. Bromine Production by Country**
- 2. Concentration by Source**
- 3. Bromine Production Capacity by Region**
- 4. Bromine Production Flow**
- 5. Bromine Compounds Market by Major Application**
- 6. Bromine Flame Retardants**
- 7. Bromine Outlook**





Bromine: the essentials

Brief description: bromine is the only liquid nonmetallic element. It is a member of the halogen group. It is a heavy, volatile, mobile, dangerous reddish-brown liquid. The red vapor has a strong unpleasant odor and the vapor irritates the eyes and throat. It is a bleaching. When spilled on the skin it produces painful sores. It is a serious health hazard, and maximum safety precautions should be taken when handling it.

Table: basic information about and classifications of bromine.

Name: Bromine

Symbol: Br

Atomic number: 35

Atomic weight: 79.904 (1)

Standard state: liquid at 298K

CAS Registry ID: 7726-95-6





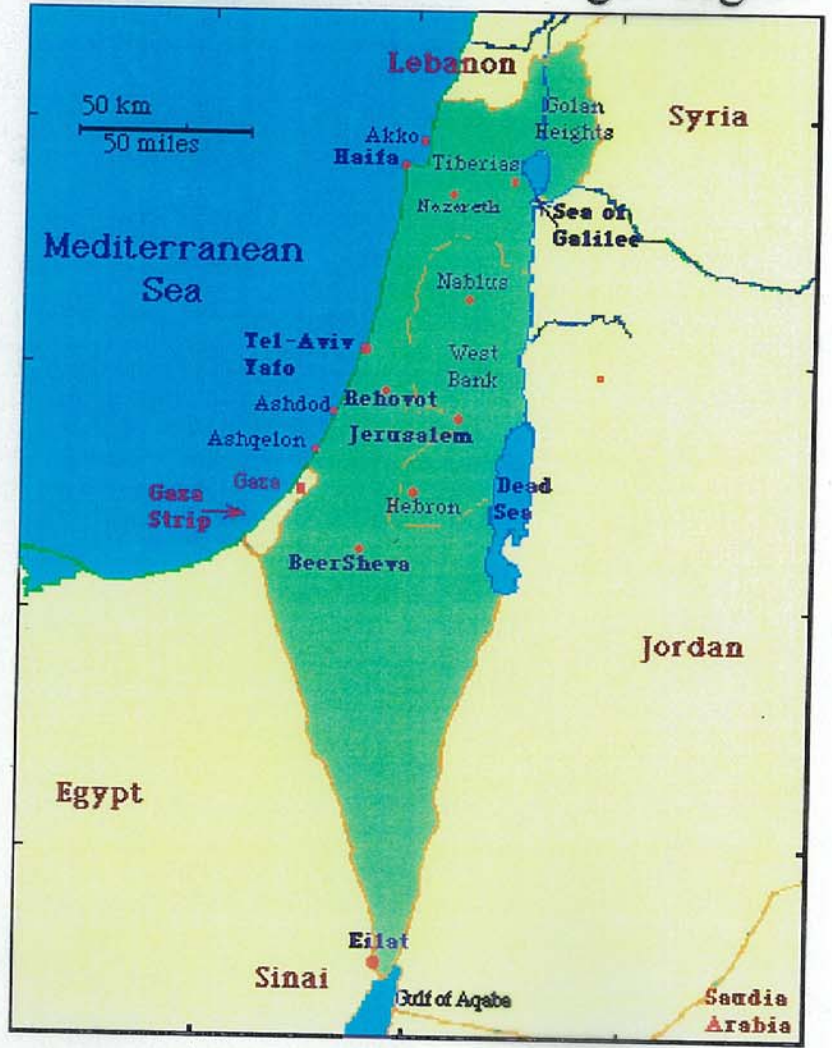
Isolation

- Isolation
- **Isolation:** bromine is available commercially so it is not normally necessary to make it in the laboratory. Bromine also occurs in seawater as the sodium salt but in much smaller quantities than chloride. It is recovered commercially through the treatment of seawater with chlorine gas and flushing through with air. In this treatment, bromide is oxidized to bromine by the chlorine gas. The principle of oxidation of bromide to bromine is shown by the addition of a little chlorine water to aqueous solutions of bromide. These become brown as elemental bromine forms.
- $2\text{Br}^- + \text{Cl}_2 \rightarrow 2\text{Cl}^- + \text{Br}_2$
- Small amounts of bromine can also be made through the reaction of solid sodium bromide, NaBr, with concentrated sulphuric acid, H₂SO₄. The first stage is formation of HBr, which is a gas, but under the reaction conditions some of the HBr is oxidized by further H₂SO₄ to form bromine and sulphur dioxide. This reaction does not work with the corresponding chlorides and fluorides.
- $\text{NaBr (s)} + \text{H}_2\text{SO}_4 \text{ (l)} \rightarrow \text{HBr (g)} + \text{NaHSO}_4 \text{ (s)}$
- $2\text{HBr (g)} + \text{H}_2\text{SO}_4 \text{ (l)} \rightarrow \text{Br}_2 \text{ (g)} + \text{SO}_2 \text{ (g)} + 2\text{H}_2\text{O (l)}$





Israel: views from the Negev region



May 1999



The Dead Sea



Dead Sea Solar evaporation ponds



View of Dead Sea area
Landsat TM3.2.1 (R.G.B)

docugraph
Produced by INFORMATION SYSTEMS



Raw material harvester





DSBG



Bromine Compounds Plant - Netherlands



Bromine/Chlorine Plant - Israel

An Introductory Presentation

23/11/00





BROMINE ISOTANKS ROUNDTRIP IMPROVEMENT



26/12/2007 16:01





Where Has the Market Share Gone? To China...



Aerial photo of Bromine production area in Shandong



Chinese Bromine production and packing facility





Bromine Production - India



Indian evaporation pond

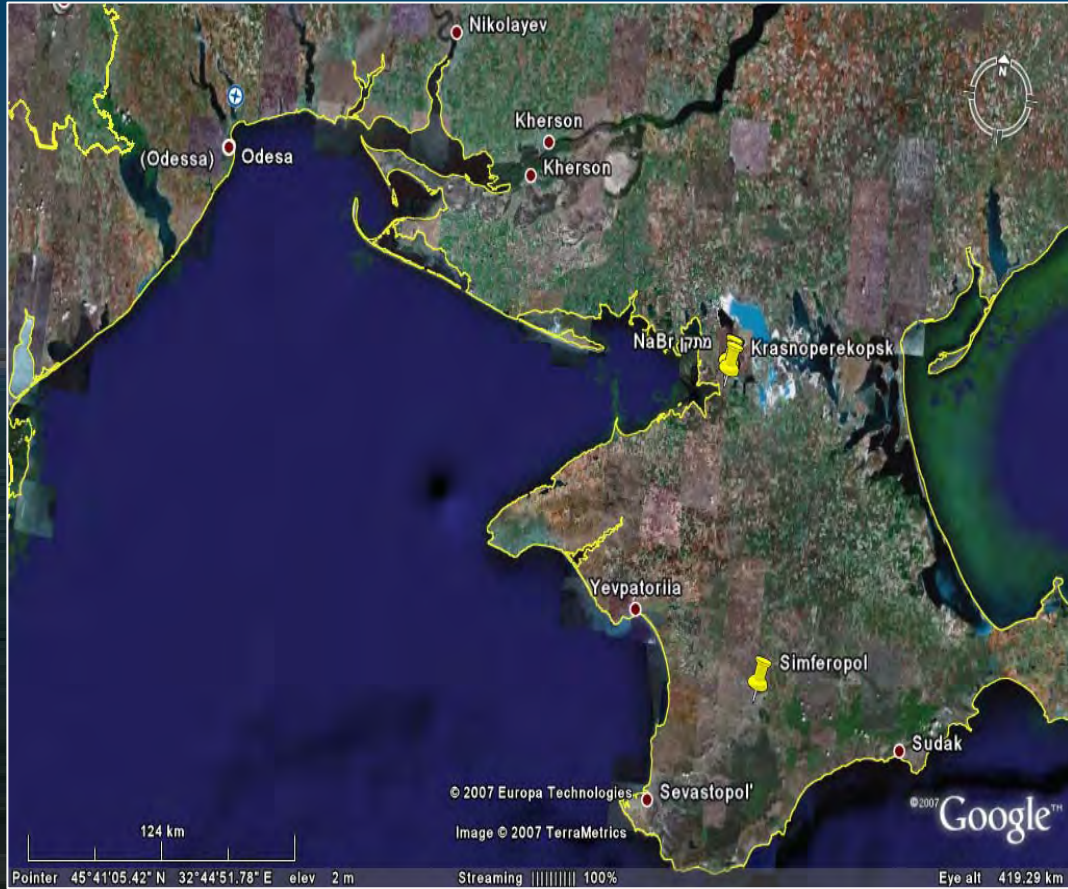


Indian plant





Emerging Ukraine ???



Satellite photo of relevant Ukrainian regions

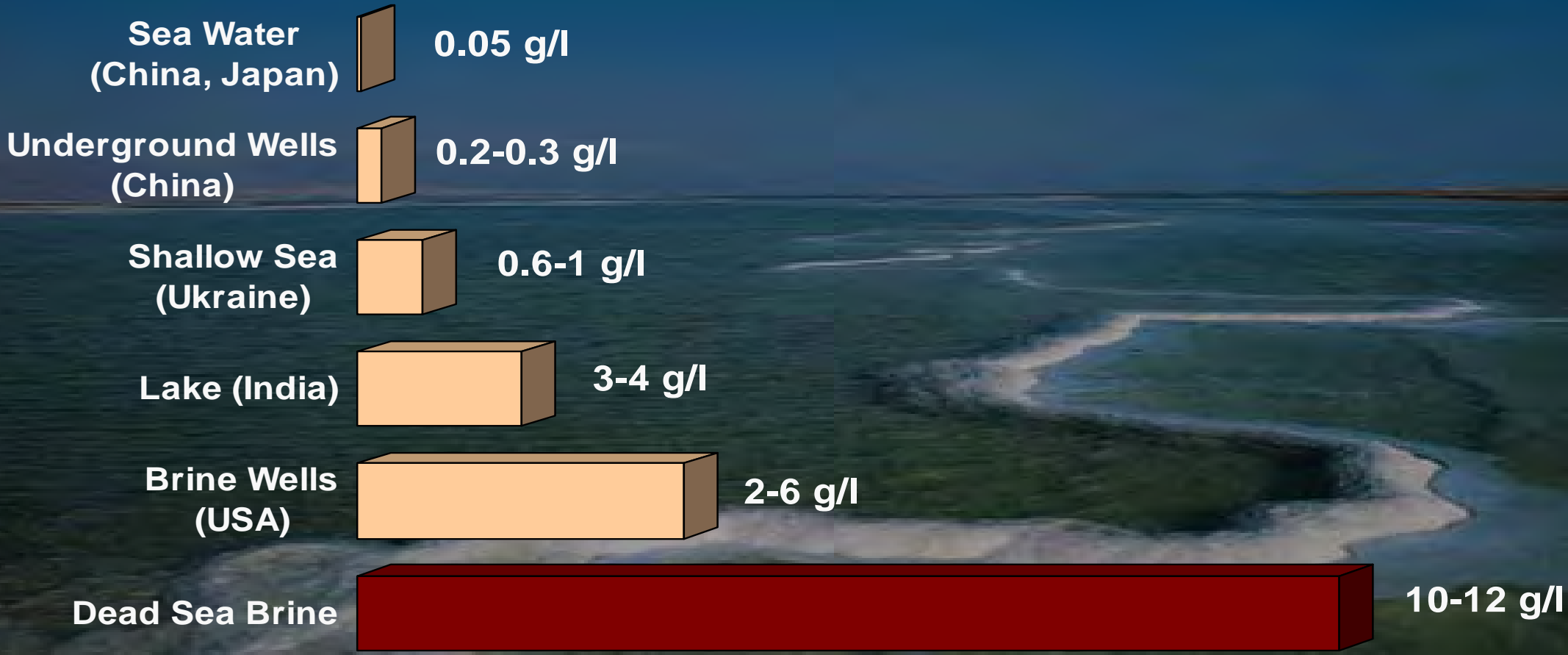


Ukrainian bromine plant





Bromine Concentration by Source

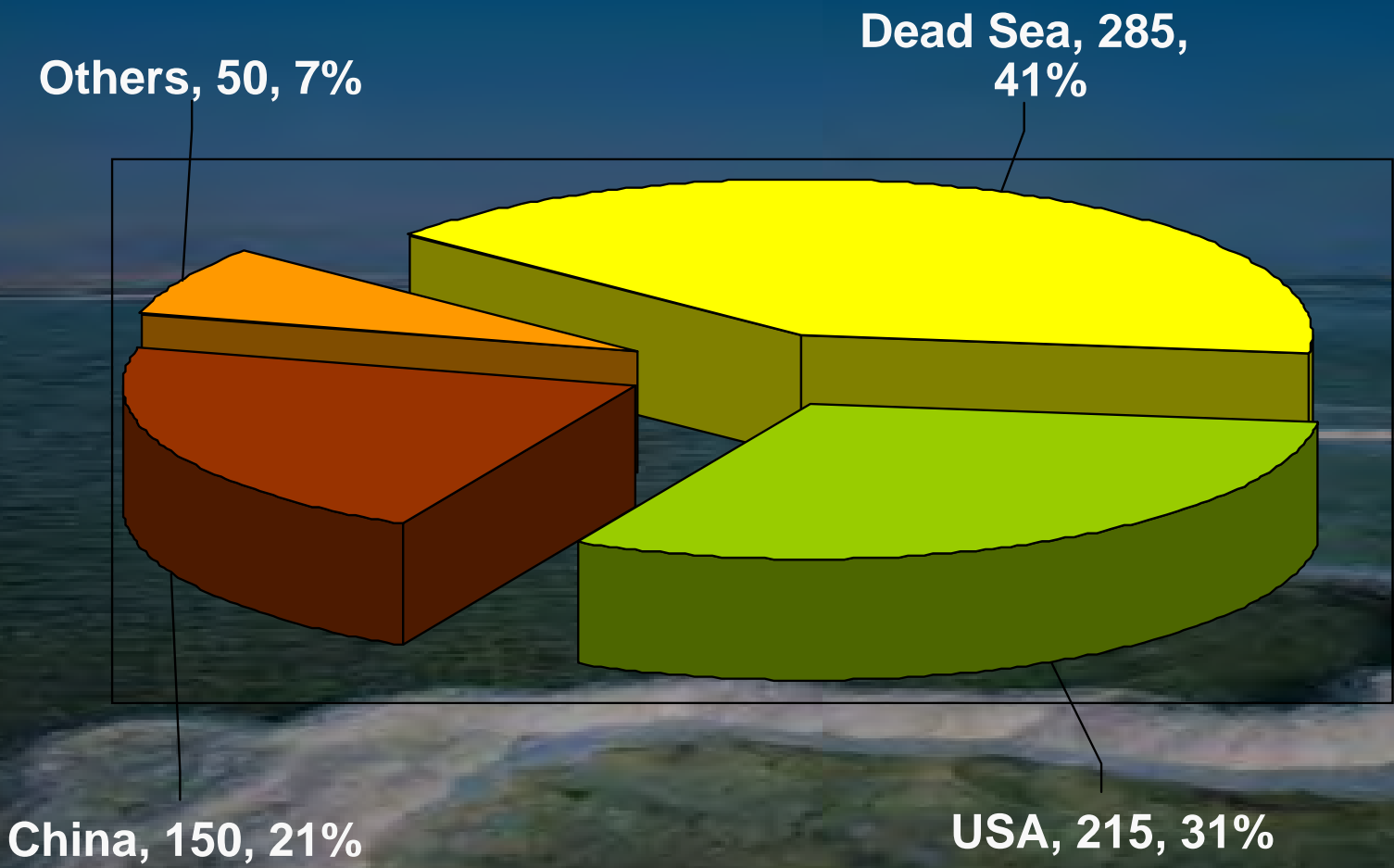


Source: ICL estimates



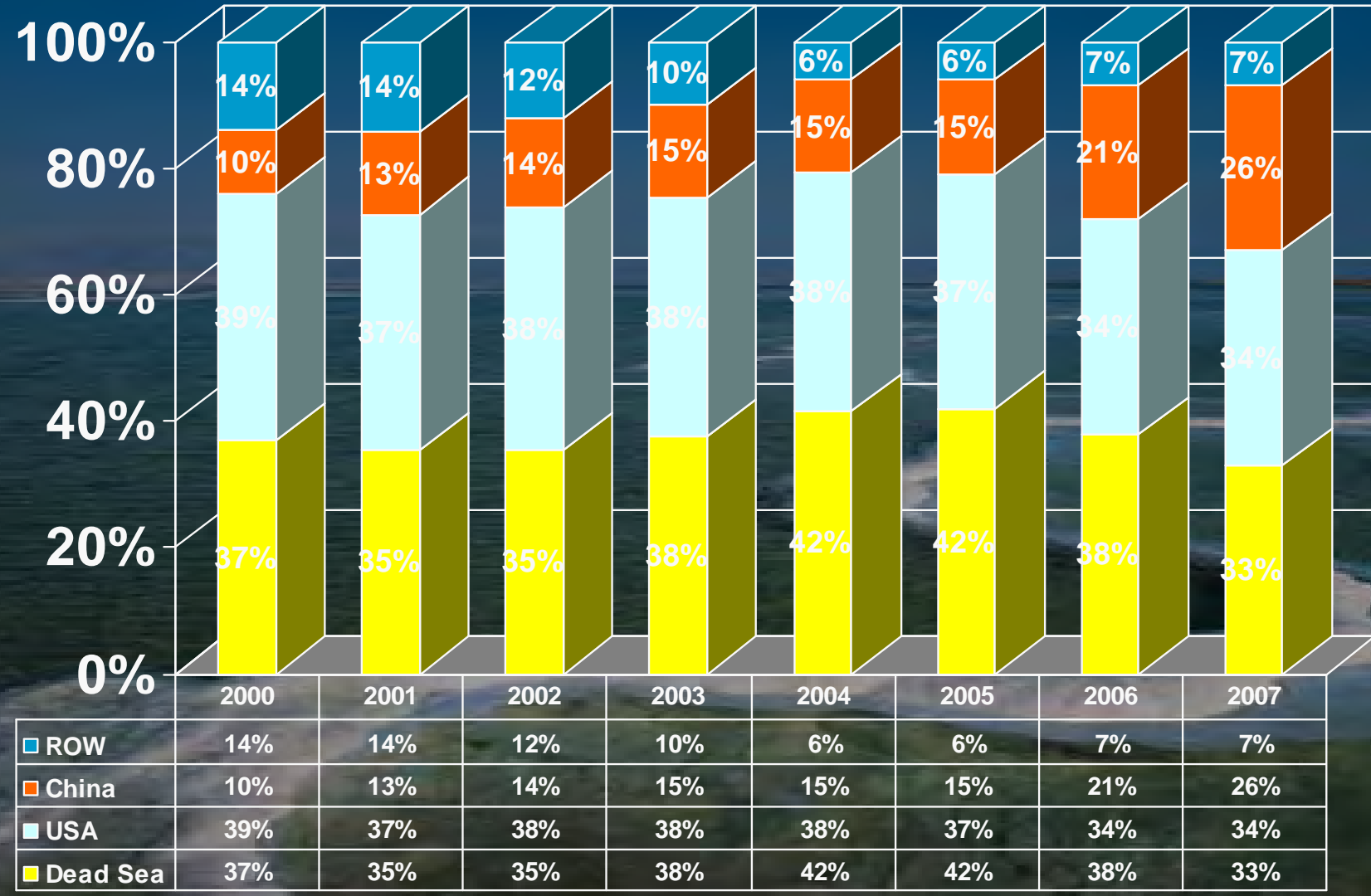


Global Bromine Production Capacity- by Region (700 KMT)





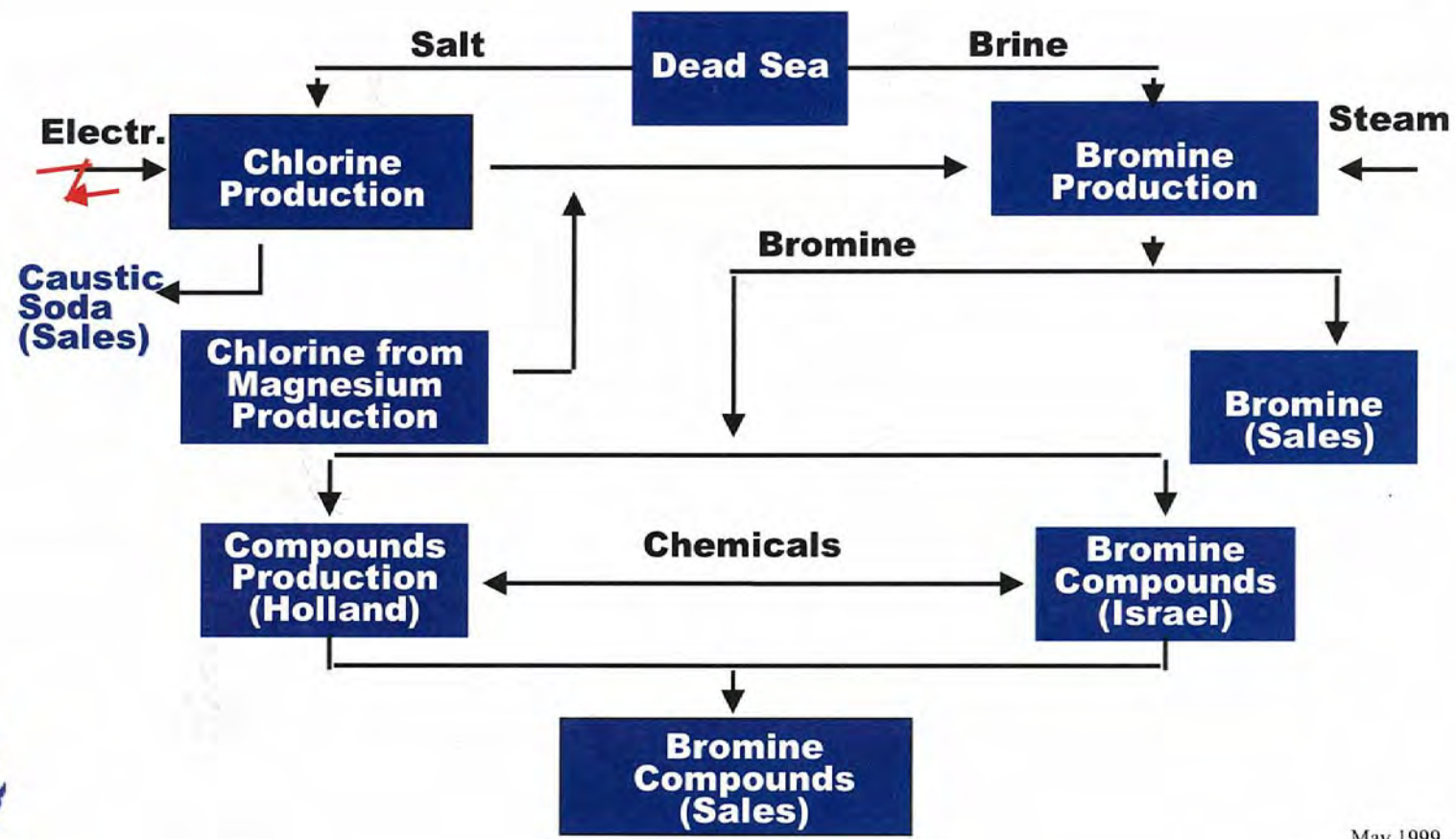
Bromine Production (by Region - % of total production)





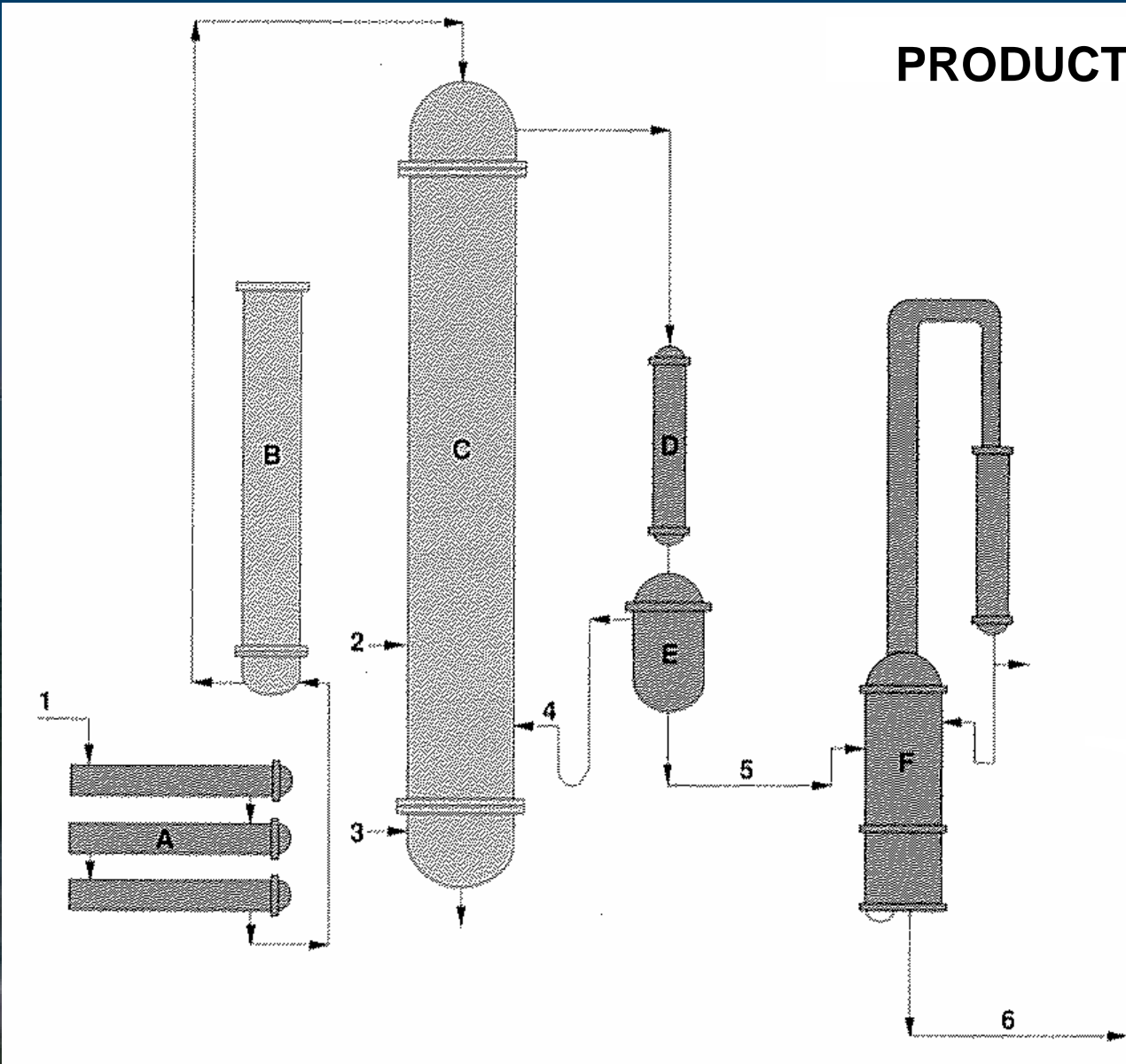
Production Flow in ICL-IP

Production Flow



May 1999





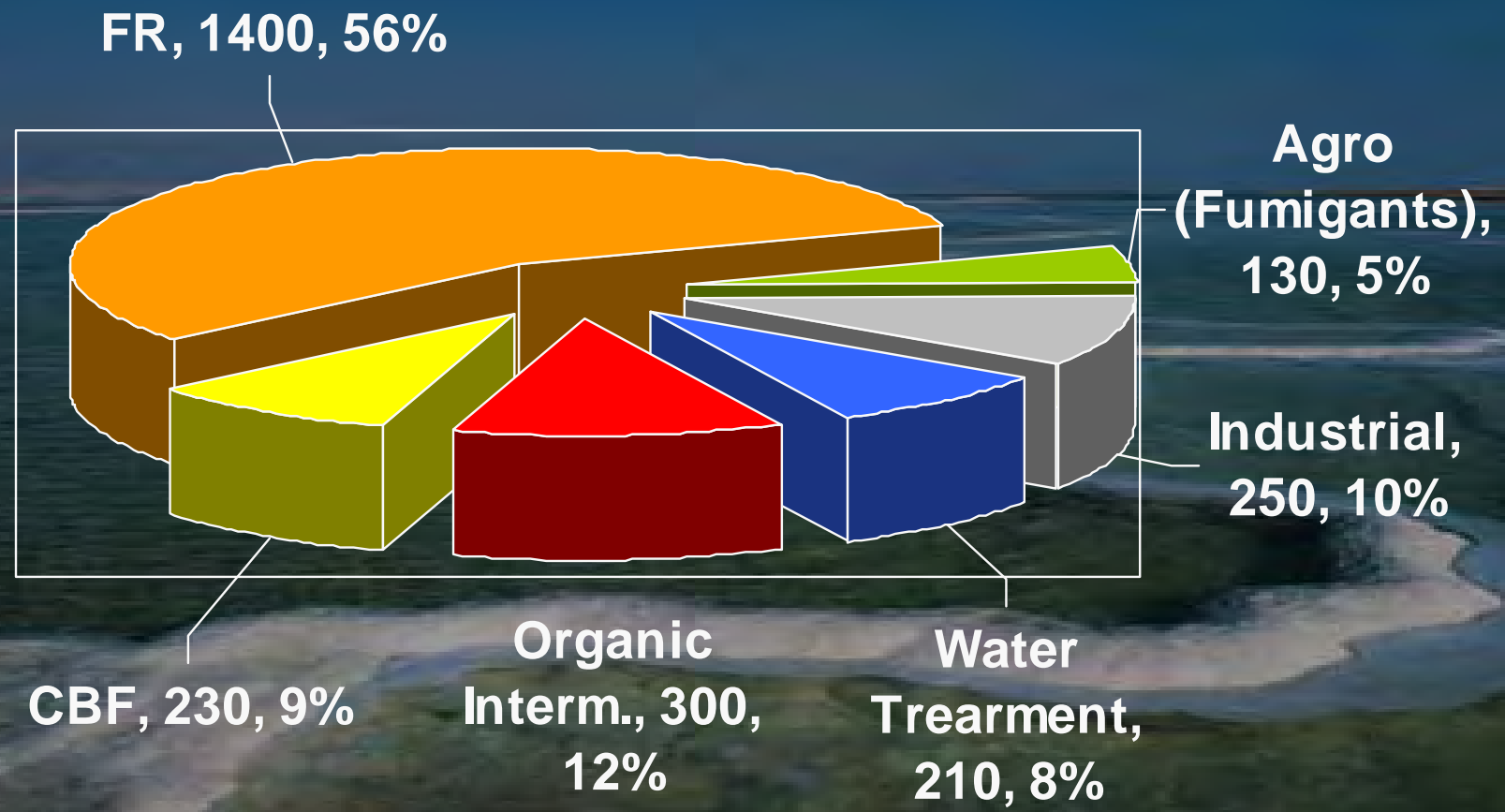
PRODUCTION FLOW CHART

LEGEND: Bromide-enriched Dead Sea brine (1), heated by preheaters (A) and heat-exchanger (B), flows into a reaction column (C) in which the introduced chlorine (2) liberated elemental bromine. Steam (3) is used to strip the bromine out of the reaction column and into a condenser (D), from which the bromine flows into a phase separator (E). The bromine-saturated water (4) is returned to the reaction column, while the bromine (5) passes, first, to a distillation column (F). After cooling, the purified liquid bromine flows to storage (6),



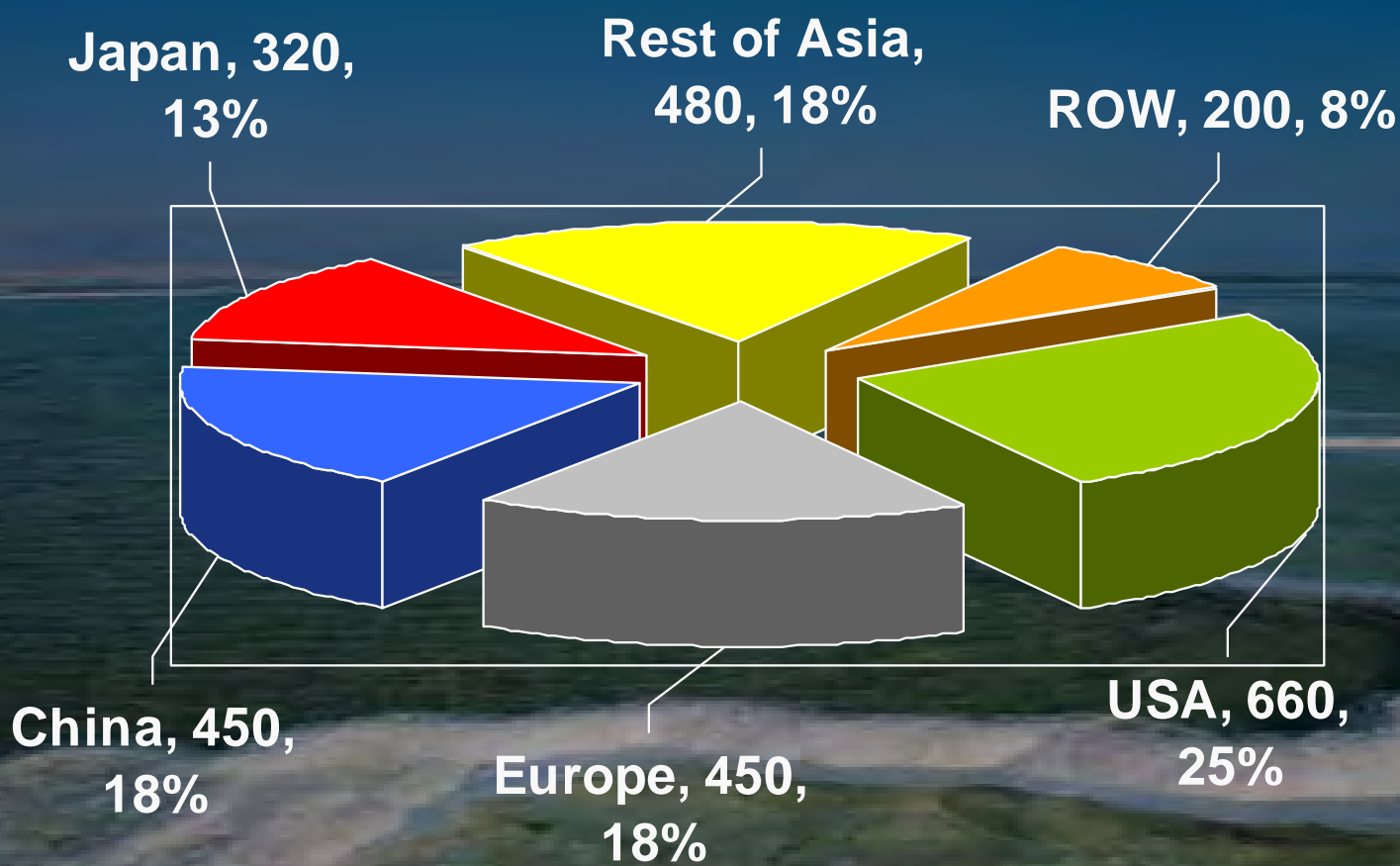


Global Bromine Compounds Market- by Major Application (Total in 2007- \$2.5 BN)



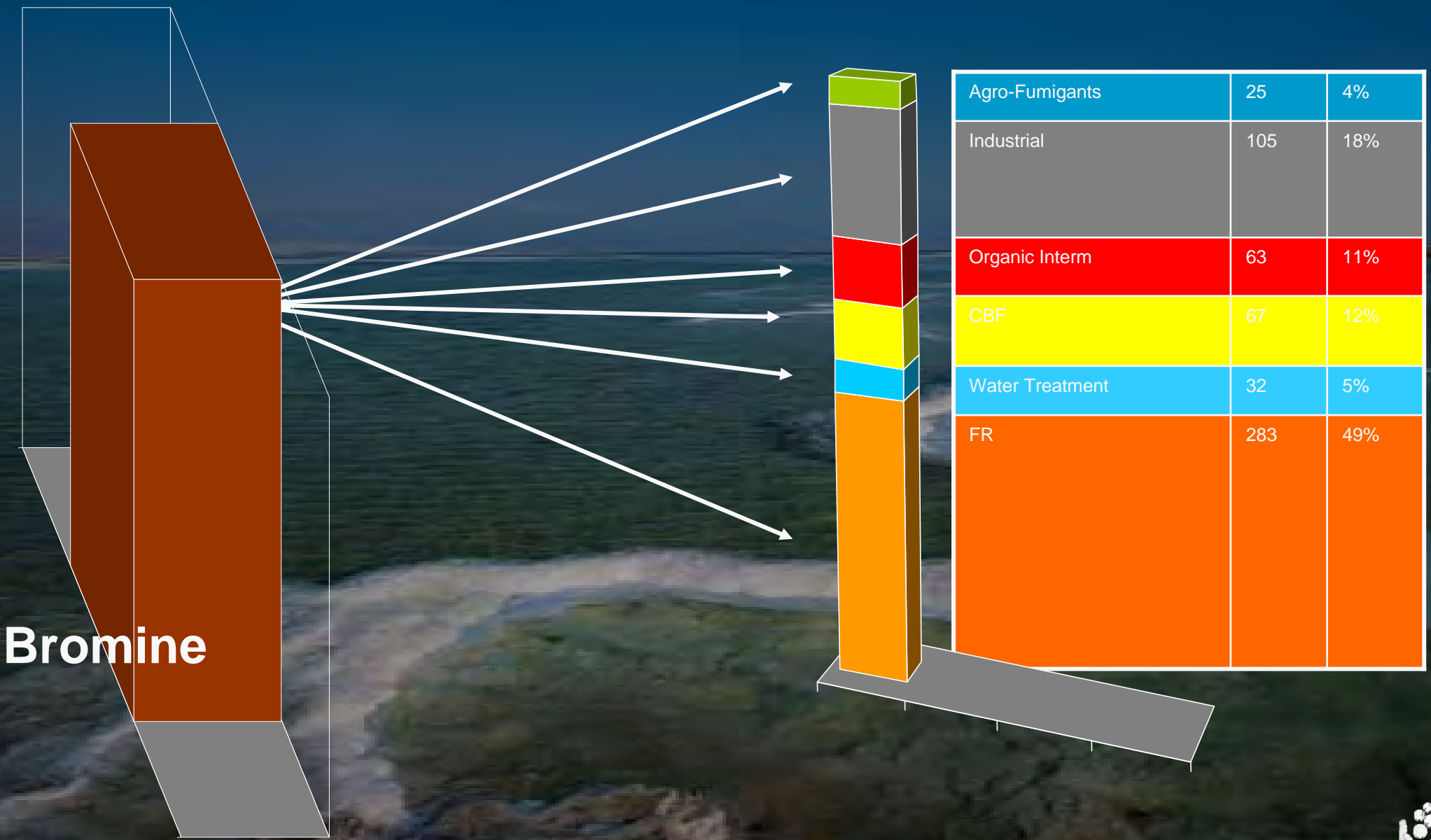


Bromine Compounds Consumption – By Region (\$2.5 BN - 2007)





Demand by Application (In terms of Bromine – 575 KMT in 2007)



Bromine





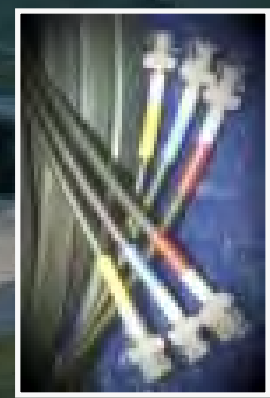
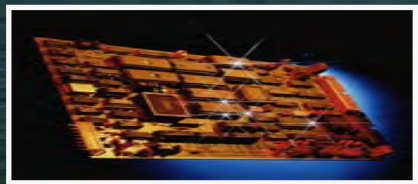
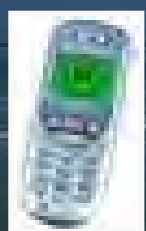
Use of Bromine Compounds

- Flame retardants
- Agrochemicals
- Water treatment chemicals
- Pharmaceuticals and cosmetics
- Lead scavenger in anti-knock fuels
- Oilfield completion fluids
- Photographic chemicals
- Desiccants for cooling systems
- Bleaching and oxidizing chemicals
- Dyestuffs
- Monomers for specialty polymers
- Precision cleaning





Electrical and electronic continues to improve the quality of our lives



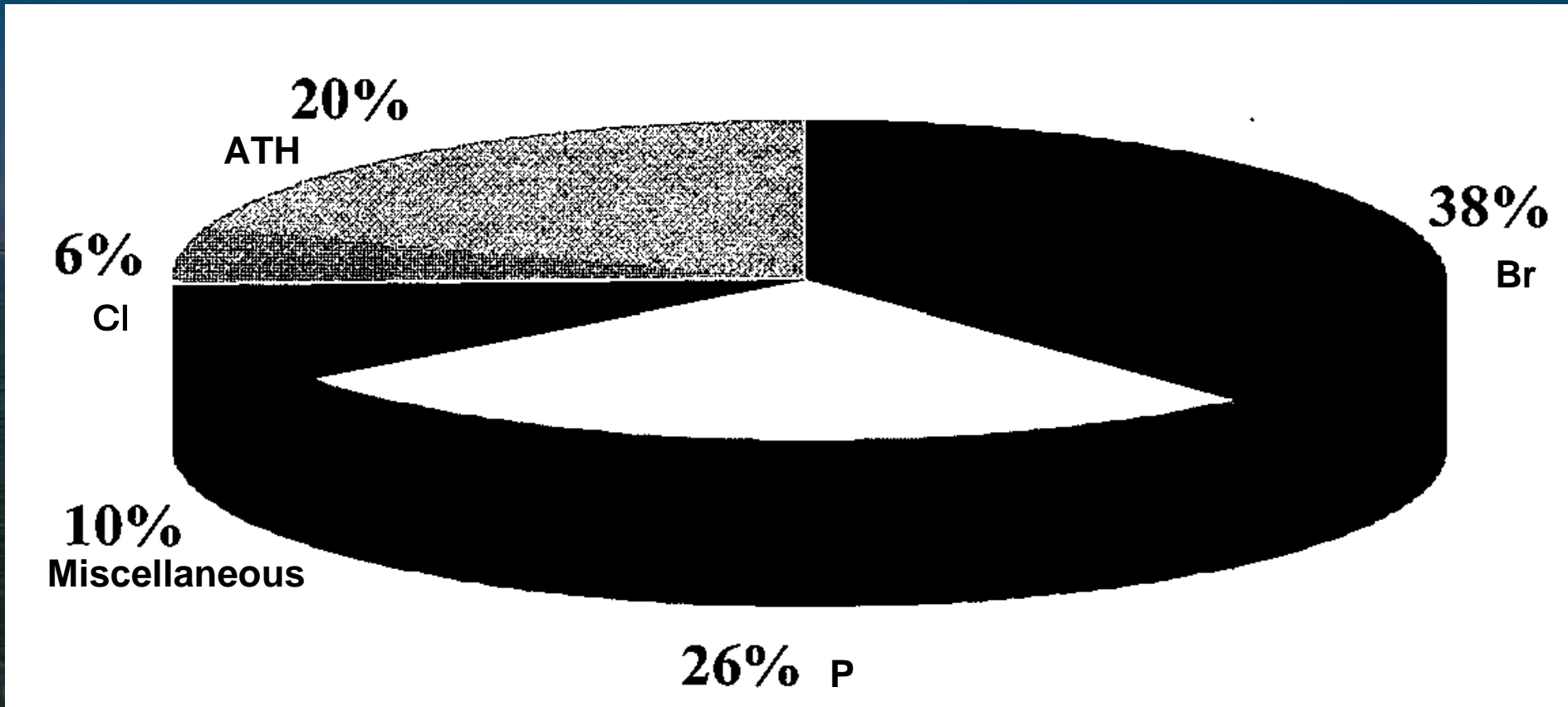
Containing Flame Retardants





Flame Retardants Worldwide Market

2,700 MM\$ in 2005 – Bromine 38%



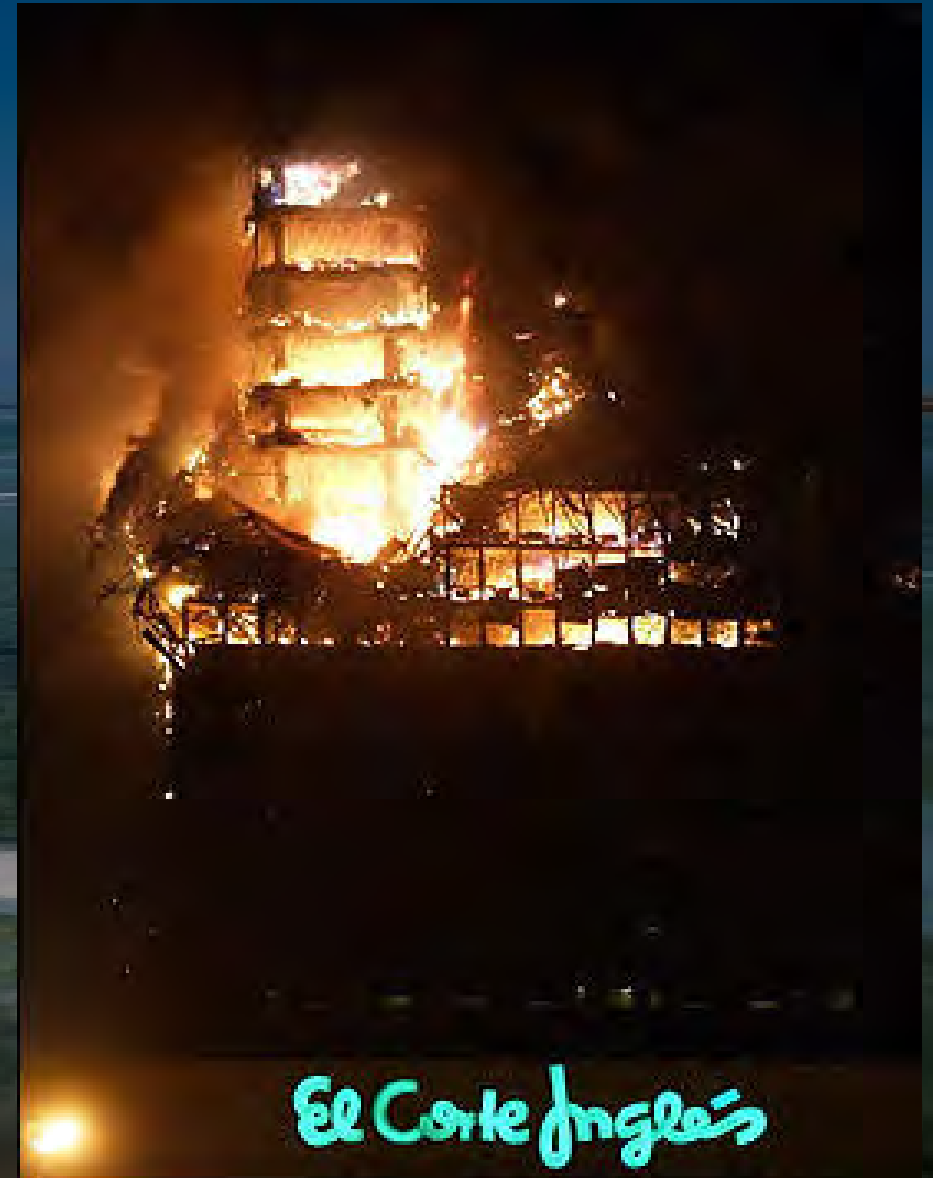
Miscellaneous include : inorganic phosphates, melamine, $Mg(OH)_2$, boron and Molybdenum compounds





February 14th 2005

100m high Windsor Tower in Madrid catching fire and completely destroyed.





The Background to VECAP Why use Flame Retardants?

Every day in Europe :

12 fire victims and 120 severely injured
75 % of victims are in their homes

Economic damage : +/- 25 billion Euro's
per year

In the UK since 1988 furniture regulation:
3200 lives saved 1988-2000
29000 less non-fatal injuries

Flame retardants and other
technologies are a critical part
of the fire safety solution.



The fire on an old (1980's)
European TV self-extinguishes,
while the fire in a non fire-safe
TV, recently sold in Europe,
keeps burning and growing





German Dioxin Ordinance , 16.7.1999

2, 3, 7, 8, - TBDD

1, 2, 3, 7, 8, - PeBDD

2, 3, 7, 8, - TBDF

2, 3, 4, 7, 8, - PeBDF 4 together : 1 ppb max.

Above 4 plus below 4 = Total 8 : 5 ppb max.

1, 2, 3, 4, 7, 8, - HxBDD

1, 2, 3, 7, 8, 9, - HxBDD

1, 2, 3, 6, 7, 8, - HxBDD

1, 2, 3, 7, 8, - PeBDD





Br-FR by RESIN APPLICATION IN JAPAN

RESIN	Br-FR
ABS	Deca, TBBA, Br-PC Olygomer, TAIC-6B, BEO, Br-PS, Ethylene-Bis(Tetrabromphtalimide) Decabromodiphenylethane, Tris(TBNPA)phosphate Br-Aromatic Triazine
EPS	TBBA-Bis(Allylether), HBCD TBP-Allylether, Tetrabromethane HIPS Deca, TBBA, HBCD, Ethylene-Bis(Tetrabromphtalimede) BEO, Tris(TBNPA)phosphate Decabromodiphenylethane Br-Aromatic Triazine TBBA-Bis(2,3Dibrompropylether)





PP TBBA-Bis(2,3Dibrompropylether)
Deca,TBBS,HBCD
TBBS-Bis(2,3Dibrompropylether)
Decabromdiphenylethane
Tris(TBNPA)phosphate
Br-Aromatic Triazine

PE Deca,
Decabromdiphenylethane
TBBS,TAIC-6B
TBBS-Bis(2,3Dibrompropylether)

PC TBBA, Br-PC Olygomer,BEO

PMMA Tris(TBNPA)phosphate
PBB-MA,PBB-PA
TAIC-6B, BEO
TPB-EO-Acrylate





PBT Deca,TBBA-2EO,TBBA-4EO
Br-PC Olygomer,PBB-PA
BEO
Ethylene-bis(Tetrabromphtalimide)
Decabromodiphenylethane

PET Deca,Br-PC Olygomer,BEO
Br-PS,TAIC-6B
Ethylene-Bis(Tetrabromphtalimide)
Decabromdiphenylethane

NYLON Deca,Poly-Dibromphenyleneoxide
Br-PS,BEO,Br-PC
Decabromdiphenylethane
Ethlene-Bis(Tetrabromphtalimide)

PVC Deca

EPDM Deca,Decabromdiphenylethane





EPOXY (laminate,BEO,)

**Deca,HBB,TBBA,TBP,BEO,
Br Aromatic Triazine
Dibromcrezylglycidylether
Dibromphenylglycidylether**

Phenol

TBBA,TBP,BEO

URETHANE FOAM

DBNPG,TBNPA,TBBA-2EO

URETHANE ELASTOMER

Deca,BEO

UPET

TBBA-2EO,DBNPG,TBNPA BEO

DCPD

Deca,Br-PS

Silicone (Rubber)

Deca,Decabromdiphenlyethane





TEXTILE (Acrylic,SBR,PP,Polyester,Cotton)

**Deca,HBCD,Pentabromtoluene
TBBA-2EO
Tris(TBNPA)phosphate**

**Adhesive,Adhesive Tape,Paper,Paint,Ink
Plywood**

**Deca,TBBA-2EO,HBCD
Ammoniumbromide
Tris(TBNPA)phosphate
Decabromdiphenylethane**





Outlook

The demand for bromine based products will grow at an average rate of 2-2.5% per year (2% in the first 3 years and later 2.7%)

➤ Major drivers and growth areas:

- Adoption of fire safety regulation.
- New applications (Mercury reduction,)
- Organic growth in the various applications, following growth of the global economy.

➤ Offset by:

- Continued phased-out of MBr
- Decline of Deca and HBCD
- Emergence of non-halogen substitutes for BFR's.





Reference

1. ICL-IP Slides, Estimated by ICL-IP
2. ICL-IP HP
3. ICL-IP Catalogue
4. VECAP Eco-Design Presentation, Hirayama

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Thank You