

**pH Calculator**To Calculate Enter Concentration:  M Enter Chemical Name/Formula: 

Input interpretation:

Result:

**4.66**

Acid-base information:

K <sub>a</sub>	$4.69 \times 10^{-11}$
pK <sub>a</sub>	10.3
pH	4.66
[H <sub>3</sub> O <sup>+</sup> ]	$2.17 \times 10^{-5}$ mol/L (moles per liter)
pOH	9.34
[OH <sup>-</sup> ]	$4.62 \times 10^{-10}$ mol/L (moles per liter)
% ionization	$2.17 \times 10^{-4}\%$

...ing...

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