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Denatured alcohol

Denatured alcohol (also called methylated spirits, in Australia, Canada, Ireland, New Zealand, South Africa, and the United Kingdom; wood spirit; and denatured rectified spirit) is ethanol that has additives to make it poisonous, bad-tasting, foul-smelling, or nauseating to discourage its recreational consumption. It is sometimes dyed so that it can be identified visually. Pyridine and methanol, [1] each and together, make denatured alcohol poisonous; and denatonium makes it bitter.

Denatured alcohol is used as a <u>solvent</u> and as fuel for <u>alcohol burners</u> and <u>camping stoves</u>. Because of the diversity of industrial uses for denatured alcohol, hundreds of additives and denaturing methods have been used. The main additive usually is 10% methanol (methyl alcohol), hence the name *methylated spirits*. Other common additives include <u>isopropyl</u> alcohol, acetone, methyl ethyl ketone, and methyl isobutyl ketone.

In the United States, mixtures sold as denatured alcohol often have less than 50% ethanol.



1 US gallon (3.8 I) of denatured alcohol in a metal container

Denaturing alcohol does not alter the ethanol molecule (chemically or structurally), unlike <u>denaturation</u> in biochemistry. Rather, the ethanol is <u>mixed with other chemicals</u> to form a foul-tasting, often toxic, solution. For many of these solutions, it is intentionally difficult to separate the components.

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Uses

In many countries, sales of alcoholic beverages are heavily <u>taxed</u> for revenue and public health policy purposes (see <u>Pigovian tax</u>). In order to avoid paying beverage taxes on alcohol that is not meant to be consumed, the alcohol must be "denatured", or treated with added chemicals to make it unpalatable. Its composition is tightly defined by government regulations in countries that tax alcoholic beverages. Denatured alcohol is used identically to ethanol itself except for applications that involve fuel, surgical

and laboratory stock. Pure ethanol is required for food and beverage applications and certain chemical reactions where the denaturant would interfere. In <u>molecular biology</u>, denatured ethanol should not be used for the <u>precipitation</u> of <u>nucleic acids</u>, since the additives may interfere with downstream applications. [2]

Denatured alcohol has no advantages for any purpose over normal ethanol; it is a public policy compromise. As denatured alcohol is sold without the often heavy taxes on alcohol suitable for consumption, it is a cheaper solution for most uses that do not involve drinking. If pure ethanol were made cheaply available for fuel, solvents, or medicinal purposes, it would likely be enjoyed as a drink by many people without payment of alcoholic beverage taxes. [3]

Toxicity

Despite its poisonous content, denatured alcohol is sometimes consumed as a <u>surrogate alcohol</u>. This can result in blindness or death if it contains methanol. For instance, during the <u>Prohibition in the United States</u>, federal law required methanol in domestically manufactured industrial alcohols. From 25–27 December 1926, which was roughly at the midpoint of the "Noble Experiment" of nationwide alcohol prohibition, 31 people in New York City alone died of methanol poisoning. [4] To help prevent this, <u>denatonium</u> is often added to give the substance an extremely bitter flavour. Substances such as <u>pyridine</u> are added to give the mixture an <u>unpleasant odour</u>, and agents such as <u>syrup of ipecac</u> may also be included to induce vomiting.

New Zealand has removed methanol from its government-approved "methylated spirits" formulation. [5][6]

In the <u>USSR</u>, denatured alcohol was used as drinking alcohol surrogate, along with many other technical ethanol-containing products. This was especially common during various <u>anti-alcohol campaigns</u> initiated by the Soviet government. There is much evidence to that in both popular folklore and in literature and music. The word "denaturat" (Russian: денатурат) even gained a special symbolic meaning. Its consumption is mentioned in songs of Vladimir Vysotsky, as well as written works of Venedikt Yerofeev, Yuz Aleshkovsky, and Vyacheslav Shishkov. [7]

Formulations

Diverse additives are used to make it difficult to use <u>distillation</u> or other simple processes to reverse the denaturation. Methanol is commonly used both because its boiling point is close to that of ethanol and because it is toxic. Another typical denaturant is pyridine. Often the denatured alcohol is dyed with methyl violet. [8]

There are several grades of denatured alcohol, but in general the denaturants used are similar. As an example, the formulation for *completely denatured alcohol*, according to 2005 British regulations was as follows: [9]

Completely denatured alcohol must be made in accordance with the following formulation: with every 90 parts by volume of alcohol mix 9.5 parts by volume of wood naphtha or a substitute and 0.5 parts by volume of crude pyridine, and to the resulting mixture add mineral naphtha (petroleum oil) in the proportion of 3.75 litres to every 1000 litres of the mixture and synthetic organic dyestuff (methyl violet) in the proportion of 1.5 grams to every 1000 litres of the mixture.

The European Union agreed in February 2013 to the mutual procedures for the complete denaturing of alcohol: [10]

Per hectolitre (100 L) of absolute ethanol: 3 litres of isopropyl alcohol, 3 litres of methyl ethyl ketone and 1 gram denatonium benzoate.

Specially denatured alcohol

A **specially denatured alcohol** (**SDA**) is one of many types of denatured alcohol specified under the United States <u>Title 27</u> of the Code of Federal Regulations Section 21.151. A specially denatured alcohol is a combination of ethanol and another chemical substance, e.g., <u>ethyl acetate</u> in SDA 29, 35, and <u>35A</u>, added to render the mixture unsuitable for drinking. SDAs are often used in cosmetic products, and can also be used in chemical manufacturing, pharmaceuticals, and solvents. Another example is **SDA 40-B**, which contains <u>tert-butyl alcohol</u> and denatonium benzoate, N.F. In the United States and other countries, the use of denatured alcohol unsuitable for beverages avoids <u>excise</u> taxes on alcohol. A

See also

- Aversive agent
- Bitterant
- Isopropyl alcohol
- Rubbing alcohol
- E85

References

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External links

- 27 CFR 20, regulations relating to denatured alcohol in the United States (https://web.archive.org/web/20120221085916/http://www.access.gpo.gov/nara/cfr/waisidx 07/27cfr20 07.html)
- Specifications and licensing of methylated spirits in the United Kingdom (http://www.legislation.go v.uk/uksi/1987/2009/contents/made)
- European Community COMMISSION REGULATION (EC) No 162/2013 on the mutual recognition of procedures for the complete denaturing of alcohol for the purposes of exemption from excise duty (http://eur-lex.europa.eu/LexUriServ/LexUriServ.do?uri=OJ:L:2013:049:0055:0061:EN:PDF)
- HM Revenue and Customs: Production, distribution and use of denatured alcohol (http://customs. hmrc.gov.uk/channelsPortalWebApp/channelsPortalWebApp.portal?_nfpb=true&_pageLabel=pageExcise_ShowContent&id=HMCE_CL_000263&propertyType=document)
- "List of SDAs with denaturing chemical" (https://www.law.cornell.edu/cfr/text/27/21.151)

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