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## THC solubility

Discussion in 'Cooking with Marijuana Recipes' started by mhughes, Feb 3, 2009.

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donnachris  
New Member

well i tried it at the ten day mark. i took 1ml three times with about 45 minutes in between. the i took almost a full tablespoon, and that was after not having any pot for three full days. i got a little relief, but not like i thought that i would.

i'm gonna try it in another 3 days. this time, i'm going to take 4 tablespoons full, evap it down to two tablespoons, and hubby and i will each take a tablepoonful. I'll let you know how that goes in another 4 or 5 days. see you then.

**Have A Smokey Day!!!**

donnachris, Mar 12, 2011

#21



Kushy  
down

1956, just curious, why do you soak it for so long, isn't that just overkill? And does 99% pure iso work any faster or should I soak my ABV in it for about a month? What do you mean "Non-THC material". Do the lower CBDS not absorb in the first week or something?

Kushy, Mar 12, 2011

#22



1956  
Sr. Member

*1956, just curious, why do you soak it for so long, isn't that just overkill? And does 99% pure iso work any faster or should I soak my ABV in it for about a month? What do you mean "Non-THC material". Do the lower CBDS not absorb in the first week or something?*

To the best of my knowledge iso alcohol is not fit for human consumption. I know people make iso hash in a matter of hours, but I have also read of people getting more than one soaking from the same weed. I just want to be sure I am giving it as much opportunity to transfer all it's properties to the alcohol.

By Non THC material I mean chlorophyll, and other plant related stuff that probably has no magic in them, but will become part of the mix none the less. Again, if I am going to smoke it to ash, might as well consume it all in a tincture

1956, Mar 12, 2011

#23



Kushy  
down

1956 said: ↑

*To the best of my knowledge iso alcohol is not fit for human consumption. I know people make iso hash in a matter of hours, but I have also read of people getting more than one soaking from the same weed. I just want to be sure I am giving it as much opportunity to transfer all it's properties to the alcohol.*

*By Non THC material I mean chlorophyll, and other plant related stuff that probably has no magic in them, but will become part of the mix none the less. Again, if I am going to smoke it to ash, might as well consume it all in a tincture*

Oh no I didn't mean making a tincture out of iso, I was planning on evaping it til it was all gone and just a pile of brown goo. So you've seen hash come out in a couple hours, but it wouldn't hurt to maybe leave the ABV iso for a day at least to get a little extra stuff?

Kushy, Mar 12, 2011

#24



1956  
Sr. Member

Buzzby wrote about putting ABV in a blender with iso to make hash this way. It was in the 420 lounge area, in like the last 4 months. It seemed like an easy way to speed the process.

1956, Mar 13, 2011

#25

1956 said: ↑



1406  
Retired

*Buzzby wrote about putting ABV in a blender with iso to make hash this way. It was in the 420 lounge area, in like the last 4 months. It seemed like an easy way to speed the process.*

I am not sure I like the idea of putting flammable liquids in a blender. It also seems redundant to 'blend' ABV bud - at least mine is usually finely ground as is. I always make hash out of my ABV, but I just shake it up in a mason jar for an hour or two, pour it onto a glass baking pan, and let the alcohol evaporate. Voila.

1406, Mar 13, 2011

#26



chemengineer  
New Member

### Easier Explanation

mhughes said: ↑

*Alrite, so we have THC is soluble at 20.2 mg THC / ml alcohol (70%)  
Also, there is approx 30ml in a fl oz*

*Thus you can put 606 mg (20.2 \* 30) of THC in 1 fl oz of 70% ethanol*

*So now we need to figure out how many grams of bud will give us  
606 mg THC.*

*X mg of bud \* 10%THC = 606mgTHC gives 6060mg of bud or  
~6grams*

*[Click to expand...](#)*

Hugh, after my computation, i conclude that you are just a bit off with your math; but you are very close. Firstly, we need to denote that 1fl ounce US is 29.6ml (set to 3 sig figs). So next we see that this question is double parted, asking to describe the function, and then compute the amount of actual grams of product (10% THC cannabis sativa needed). The correct function describing this is such:  $(20.2\text{mg THC}/1\text{ml ethyl})(X \text{ ounces of ethyl alcohol} \times 29.6\text{ml} \times .70)(10)$  This is roughly formated bc of the threat parameter. Anyway, in 1 ounce of 70% ethyl alcohol (ethyl is a alkene functional group, and is noted as X% because ethyl alcohol is only X% ethyl functional group, in simple terms.) 4185.44mg of actual product is needed. This simply converts to about 4.2 grams needs per 1 ounce of 70% ethyl alcohol. Im not sure why hugh's math was off but maybe it was the awkward wording of the question itself. A few chemical principles we have to remember are: aqueous solubility is denoted as g/ml. Specifically grams solute over ml solvent. Solubility always involves 2 different compounds/ solutions, hence the reason for its units. Another thing we have to remember is to word questions properly and more importantly, correctly decipher questions presented. this math is very basic and trivial, but i oblige to show you the derivation of each step, and the chemical mechanisms that occur when this reaction occurs. Anyway, dont mean to be verbose or in use of complicated nomenclature, but chemistry is an art and there is only ONE correct answer to math based problems like such. Thanks and feel free to debate. -Joe

chemengineer, Jul 17, 2011

#27



**i explained the correct version and the function associated with such**

chemengineer, Jul 17, 2011

#28

chemengineer  
New Member

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